

# Summary of Product Characteristics

## 1 NAME OF THE MEDICINAL PRODUCT

Bisodol Antacid Chewable Tablets Calcium Carbonate 522mg Magnesium Carbonate Light 68mg Sodium Hydrogen Carbonate 64mg

## 2 QUALITATIVE AND QUANTITATIVE COMPOSITION

Each chewable tablet contains:

64 mg of Sodium Hydrogen Carbonate  
522 mg of Calcium Carbonate  
68 mg of Magnesium Carbonate, Light

### Excipients with known effect:

Each chewable tablet also contains 538.3mg of sucrose  
Each chewable tablets also contains 18mg of sodium

For a full list of excipients, see section 6.1.

## 3 PHARMACEUTICAL FORM

Chewable Tablet

Circular, smooth white bevel-edged tablet marked Bisodol on both sides.

## 4 CLINICAL PARTICULARS

### 4.1 Therapeutic Indications

In the relief of symptoms of indigestion, dyspepsia, heartburn, acidity and flatulence.

### 4.2 Posology and method of administration

#### Posology

##### *Adults*

Two tablets as required.

##### *Paediatric population*

Children under 12 years: not recommended

Do not take more than 12 tablets in 24 hours.

#### Method of administration

Bisodol Chewable Tablets are to be administered orally

### 4.3 Contraindications

Hypersensitivity to the active substances or to any of the excipients listed in section 6.1.

Hypophosphataemia. Avoid in patients with renal failure.

Hypercalcaemia and hypercalciuria.

#### 4.4 Special warnings and precautions for use

- Magnesium salts may cause central nervous depression in the presence of renal insufficiency.
- Persons with kidney disease or receiving medical treatment should consult their doctor before using.
- Prolonged use should be avoided. If the symptoms persist, medical advice should be sought.
- Do not exceed the stated dose except on medical advice.
- Ingestion of large amounts of antacid tablets may cause milk-alkali syndrome.
- Bisodol contains sucrose: Patients with rare hereditary problems of fructose intolerance, glucose-galactose malabsorption or sucrase-isomaltase insufficiency should not take this medicine.

#### 4.5 Interaction with other medicinal products and other forms of interactions

As with other antacids, Bisodol Antacid Chewable Tablets may form complexes with certain drugs e.g. tetracyclines, digoxin and vitamins resulting in decreased absorption. Bisodol Antacid Chewable Tablets should not be administered within two hours of taking such drugs. Bisodol Antacid Chewable Tablets can also impede the absorption of phosphates.

#### 4.6 Fertility, pregnancy and lactation

Animal studies are insufficient with respect to effects on pregnancy/embryonal/foetal development/parturition and postnatal development.

Caution should be exercised when prescribing to pregnant women.

#### 4.7 Effects on ability to drive and use machines

None stated.

#### 4.8 Undesirable effects

Calcium salts can have a constipating effect and magnesium salts can have a laxative effect.

The specific mixture of antacids is intended to avoid the lower gastrointestinal effects seen with a single antacid preparation. No side effects are associated with sodium bicarbonate except when taken in excess. Rebound hyperacidity may occur with prolonged usage.

#### Reporting of suspected adverse reactions

Reporting suspected adverse reactions after authorisation of the medicinal product is important. It allows continued monitoring of the benefit/risk balance of the medicinal product. Healthcare professionals are asked to report any suspected adverse reactions via HPRC Pharmacovigilance Website: [www.hpra.ie](http://www.hpra.ie).

#### 4.9 Overdose

Chronic intake can lead to hypercalcaemia, hypercalciuria, nephrolithiasis, metabolic alkalosis and renal insufficiency. Acid rebound may also occur.

### 5 PHARMACOLOGICAL PROPERTIES

#### 5.1 Pharmacodynamic properties

Sodium bicarbonate, calcium carbonate and magnesium carbonate are antacids. They act by neutralising the hydrochloric acid produced by the stomach and thus reducing gastric and duodenal irritation.

#### 5.2 Pharmacokinetic properties

##### Calcium Carbonate

Calcium carbonate is converted to calcium chloride by gastric acid. Some of the calcium is absorbed from the intestines but about 85% is reconverted to insoluble calcium salts such as the carbonate and is excreted in the faeces.

## **Magnesium Carbonate**

Magnesium carbonate reacts with gastric acid to form soluble magnesium chloride and carbon dioxide in the stomach. Some magnesium is absorbed but is usually excreted rapidly in the urine.

## **Sodium Bicarbonate**

Administration of sodium bicarbonate by mouth causes neutralisation of gastric acid with the production of carbon dioxide. Bicarbonate not involved in that reaction is absorbed and in the absence of a deficit of bicarbonate in the plasma, bicarbonate ions are excreted in the urine, which is rendered alkaline with an accompanying diuresis.

### **5.3 Preclinical safety data**

None stated

## **6 PHARMACEUTICAL PARTICULARS**

### **6.1 List of excipients**

Saccharin soluble  
Maize starch  
Sucrose  
Calcium stearate  
Peppermint Essential Oil

### **6.2 Incompatibilities**

Not applicable.

### **6.3 Shelf life**

5 years

### **6.4 Special precautions for storage**

Do not store above 25°C.

### **6.5 Nature and contents of container**

Packsizes of 100 contained in cellophane over wrapped carton of 5 rolls of 20 tablets in wax laminated foil with paper labels.

Packsizes of 30 contained in cellulose over wrapped shell and slide cartons.

Packsizes of 20 contained in wax laminated foil with paper label.

Packsizes of 30 contained in a polypropylene container with a polypropylene lid.

Not all packsizes may be marketed.

### **6.6 Special precautions for disposal and other handling**

No special requirements for disposal. Any unused medicinal product or waste material should be disposed of in accordance with local requirements.

## **7 MARKETING AUTHORISATION HOLDER**

Teva B.V.  
Swensweg 5  
2031GA Haarlem  
Netherlands

**8 MARKETING AUTHORISATION NUMBER**

PA1986/071/001

**9 DATE OF FIRST AUTHORISATION/RENEWAL OF THE AUTHORISATION**

Date of first authorisation: 01 April 1983

Date of last renewal: 01 April 2008

**10 DATE OF REVISION OF THE TEXT**

April 2022